

Lots of Ionic Naming Practice Problems - Solutions

Name the following ionic compounds:

- 1) NaBr **sodium bromide**
- 2) Sc(OH)₃ **scandium hydroxide**
- 3) V₂(SO₄)₃ **vanadium (III) sulfate**
- 4) NH₄F **ammonium fluoride**
- 5) CaCO₃ **calcium carbonate**
- 6) NiPO₄ **nickel (III) phosphate**
- 7) Li₂SO₃ **lithium sulfite**
- 8) Zn₃P₂ **zinc phosphide**
- 9) Sr(C₂H₃O₂)₂ **strontium acetate**
- 10) Cu₂O **copper (I) oxide**
- 11) Ag₃PO₄ **silver phosphate**
- 12) YClO₃ **yttrium chlorate**
- 13) SnS₂ **tin (IV) sulfide**
- 14) Ti(CN)₄ **titanium (IV) cyanide**
- 15) KMnO₄ **potassium permanganate**
- 16) Pb₃N₂ **lead (II) nitride**
- 17) CoCO₃ **cobalt (II) carbonate**
- 18) CdSO₃ **cadmium sulfite**
- 19) Cu(NO₂)₂ **copper (I) nitrite**
- 20) Fe(HCO₃)₂ **iron (II) bicarbonate**

